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% Problem 3
clear;clc;

% Define constants and parameters
R = 8.314; %Universal gas constant (kJ/molK)
temp = [1773 1673 1573 1473 1373 1273 1173 1073 1023]; % Temperatures at which data was collected.
delta = [10^-1 10^-1.5 10^-2 10^-2.5 10^-3]; % Range chosen to cover a wide span of experimental data

% Import data obtained experimentally such as that seen in Figure 22.12
dataTable = readtable('Fig2212_data.csv');

% Convert data table into an array.
% Note: you can refer to the table form for column names
dataArray = table2array(dataTable);

% Organize temperature, delta, and pO2 data into a structure and assign
% polynomial fits of pO2 = f(delta)
for i = 1:length(temp)

    num = strcat('T',num2str(temp(i)));
    data.(num)(:,1) = dataArray(:,2*i-1); % pO2 (atm)
    data.(num)(:,2) = dataArray(:,2*i); % delta

    % Use this for loop to delete NaN entries from data import caused by
    % an unequal number of data points available for each isotherm. This
    % will allow for creation of a polynomial fit.
    for j = 1:length(data.(num))
        if isnan(data.(num)(j,1)) == true
            data.(num)(j:length(data.(num)),:)=[];
            break
        end
    end

    % Create a polynomial structure and store polynomial fits.
    % Choose an order that best fits your data. Here, a fifth order
    % polynomial is used.

    % Note: the data is fitted in log10 form to help more accurately fit the
    % data.
    poly.(num) = polyfit(log10(data.(num)(:,2)),log10(data.(num)(:,1)),5);

end

% For each constant value of delta, search each set of temperature data and
% check if it contains the current constant delta value.
% If so, evaluate the polynomial for that temperature at the current constant
% delta to obtain pO2 using the polyval function.

% Store that value as 0.5*log(pO2), along with the inverse of the
% temperature at which that value was solved. Here, these values are being
% stored in a data structure called ellinghamData.

% Note:log(x) = natural logarithm in MATLAB v2020b
for i = 1:length(delta)
    k = 1;
    num1 = strcat('d',num2str(i));

    for j = 1:length(temp)
        num2 = strcat('T',num2str(temp(j)));
        if delta(i) > min(data.(num2)(:,2)) && delta(i) < max(data.(num2)(:,2))
            ellinghamData.(num1)(k,1) = 0.5*log(10^polyval(poly.(num2),log10(delta(i))));
            ellinghamData.(num1)(k,2) = 1/temp(j);
            k = k+1;
        end
    end
end

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end
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% Plot the data obtained for the ellingham diagram for each constant delta
% value. Inverse temperature will go on the x-axis, while  $0.5\ln(p_{O_2})$  will
% go on the y-axis. Fit a line to each constant delta data set and plot.
% Scale the slope and intercept of the linear fit for each delta
% appropriately and store each in an array.
for i = 1:length(delta)

    col = rand(3,1);
    num = strcat('d',num2str(i));
    plot(ellinghamData.(num)(:,2),ellinghamData.(num)(:,1), 'o', 'Color', col, 'HandleVisibility', 'Off');
    hold on;
    polyProp.(num) = polyfit(ellinghamData.(num)(:,2),ellinghamData.(num)(:,1),1);
    plot(1./temp,polyval(polyProp.(num),1./temp), '-', 'Color', col);

    H(i) = -(polyProp.(num)(1)*R)/1000;
    S(i) = (polyProp.(num)(2)*R);
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end
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% Plot labels
xlabel('1/T (K-1)');
ylabel('ln( $\frac{p_{O_2}}{p^0}$ )');
legend('\delta=10-1', '\delta=10-1.5', '\delta=10-2', '\delta=10-2.5', '\delta=10-3')

% Concatenate thermodynamic property arrays and output as a table
output = cat(1,H,S);
array2table(output, 'VariableNames', {'delta=10-1', 'delta=10-1.5', 'delta=10-2'...
    , 'd=10-2.5', 'd=10-3'}, 'RowNames', {'Enthalpy (kJ/mol)', 'Entropy (J/molK)'})
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ans =
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2x5 table
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	delta=10 ⁻¹	delta=10 ^{-1.5}	delta=10 ⁻²	d=10 ^{-2.5}	d=10 ⁻³
Enthalpy (kJ/mol)	399.411993399213	443.119281300684	472.56118350642	482.76937807391	483.917465286416
Entropy (J/molK)	164.159749819852	214.416176126595	254.027038394741	284.674518892935	308.785108716772

